organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Sibel Demir,^a Muharrem Dinçer,^a Alaaddin Çukurovalı^b and Ibrahim Yılmaz^c*

^aOndokuz Mayıs University, Arts and Sciences Faculty, Department of Physics, 55139 Samsun, Turkey, ^bFırat University, Arts and Sciences Faculty, Department of Chemistry, 23119-Elazığ, Turkey, and ^cFırat University, Arts and Sciences Faculty, Department of Chemistry, 23119 Elazığ, Turkey

Correspondence e-mail: sibeld@omu.edu.tr

Key indicators

Single-crystal X-ray study T = 296 KMean $\sigma(\text{C}-\text{C}) = 0.004 \text{ Å}$ Disorder in main residue R factor = 0.070 wR factor = 0.195 Data-to-parameter ratio = 17.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-Chloro-1-(3-mesityl-3-methylcyclobutyl)ethanone

The title compound, $C_{16}H_{21}CIO$, has a non-planar configuration. The crystal packing is mainly stabilized by intramolecular $C-H\cdots O$ hydrogen-bonding interactions. Received 1 December 2005 Accepted 12 December 2005 Online 21 December 2005

Comment

It has been shown that 3-substituted cyclobutane carboxylic acid derivatives have antidepressant activities and liquidcrystal properties (Dehmlow & Schmidt, 1990; Escaler *et al.*, 1977). Substituted α -haloketones, like the title compound, (I), are used for different purposes, especially in the synthesis of heterocyclic substances (Çukurovalı *et al.*, 2002; Gompper & Christmann, 1959). The extensive synthetic possibilities of this compound, due to the presence of various reaction sites, hold promise for the preparation of new heterocyclic chemicals.



Fig. 1 shows the molecular structure and conformation of (I) with the atomic numbering scheme. Selected bond lengths and angles in (I) are given in Table 1. In cyclobutane ring A (C1–C4), the puckering parameter Q_2 is -0.228 (2) Å (Cremer & Pople, 1975). In this study, the C4/C1/C2 plane forms a dihedral angle of 23.8 (3)° with the C2/C3/C4 plane, which deviates from the values reported in previous studies (Swenson *et al.*, 1997; Yüksektepe *et al.*, 2004). The geometry of the cyclobutane ring is influenced by the steric effect of the methyl group. The bond lengths and angles in the four-membered ring



Figure 1

An *ORTEP3* (Farrugia, 1997) drawing of (I), showing the atomic numbering scheme. Displacement ellipsoids of non-H atoms are drawn at the 30% probability level. The minor occupancy disordered atoms have been omitted.

© 2006 International Union of Crystallography Printed in Great Britain – all rights reserved



Figure 2

A projection of the molecular packing of (I) approximately along the aaxis. The minor occupancy disordered atoms have been omitted.

are normal (Allen et al., 1987). The chloroacetaldehyde group shows disorder and atoms C15, C16, O1 and Cl1 were modelled in two different orientations, with occupancy factors of 0.61 (2) for atoms with suffix A and 0.39 (2) for those with suffix B (Fig. 1).

Experimental

The synthesis of the title compound was realised according to the literature method of Akhmedov et al. (1991) with some modification, as given in the reaction scheme. Shiny crystals suitable for X-ray analysis were obtained by crystallization from ethanol (yield: 80%, m.p. 366 K). IR (KBr, ν cm⁻¹): 1724 (C=O), 732 (-CH₂-Cl). ¹H NMR (CDCl₃): δ 1.63 (s, 3H,CH₃ on cyclobutane), 2.22 (s, 3H, p-CH₃ on mesitylene), 2.25 (s, 6H, o-CH₃ on mesitylene), 2.42-2.78 (m, 4H, $-CH_{2}$ -), 6.80 (s, 2H, aromatics on mesitylene). ¹³C NMR (CDCl₆): δ 48.88 (C₁), 205.47 (C₂), 41.89 (C₃), 42.59 (C₄), 26.95 (C₅), 40.19 (C₆), 145.09 (C₇), 132.48 (C₈), 132.44 (C₉), 136.96 (C₁₀), 22.46 (C₁₁), 23.37 (C₁₂).

Crystal data

C₁₆H₂₁ClO $M_r = 264.78$ Monoclinic, $P2_1/c$ a = 8.3808 (17) Åb = 15.236 (4) Å c = 12.472 (3) Å $\beta = 112.173 (15)^{\circ}$ V = 1474.8 (6) Å³ Z = 4

Data collection

Stoe IPDS-2 diffractometer ω scans Absorption correction: integration (X-RED32; Stoe & Cie, 2002) $T_{\min} = 0.841, T_{\max} = 0.914$ 23037 measured reflections 3378 independent reflections

 $D_x = 1.192 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation Cell parameters from 26744 reflections $\theta = 2.2 - 27.9^{\circ}$ $\mu = 0.25~\mathrm{mm}^{-1}$ T = 296 KPrism, colourless 0.73 \times 0.57 \times 0.37 mm

1993 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.069$ $\theta_{\rm max} = 27.8^{\circ}$ $h = -10 \rightarrow 10$ $k = -19 \rightarrow 20$ $l = -16 \rightarrow 16$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.070$	H-atom parameters constrained $w = 1/[\sigma^2(F^2) + (0.1242P)^2]$
$vR(F^2) = 0.195$	where $P = (F_o^2 + 2F_c^2)/3$
S = 0.96	$(\Delta/\sigma)_{\rm max} < 0.001$
378 reflections	$\Delta \rho_{\rm max} = 0.25 \ {\rm e} \ {\rm A}^{-3}$
.97 parameters	$\Delta \rho_{\rm min} = -0.19 \ {\rm e} \ {\rm \AA}^{-3}$

Table 1 Selected geometric parameters (Å, °).

Cl1A - C16A $Cl1B - C16B$	1.807 (12)	O1 <i>A</i> -C15 <i>A</i>	1.187 (16)
	1.68 (2)	O1 <i>B</i> -C15 <i>B</i>	1.22 (2)
O1A - C15A - C2	121.2 (11)	O1 <i>B</i> -C15 <i>B</i> -C2	125.8 (18)
O1A - C15A - C16A	126.3 (12)	Cl1 <i>A</i> -C16 <i>A</i> -C15 <i>A</i>	113.5 (8)
O1B-C15B-C16B	115.3 (18)	Cl1B-C16B-C15B	113.8 (13)

H atoms were positioned geometrically and treated using a riding model, fixing the bond lengths at 0.93, 0.96, 0.97 and 0.98 Å for aromatic, methyl, methylene and methine H atoms, respectively. The displacement parameters of the H atoms were constrained to be $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}$ of the carrier atom.

Data collection: X-AREA (Stoe & Cie, 2002); cell refinement: X-AREA; data reduction: X-RED32 (Stoe & Cie, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON (Spek, 2003).

This study was supported financially by the Research Centre of Ondokuz Mayıs University (project No. F-276).

References

- Akhmedov, M. A., Sardarov, I. K., Akhmedov, I. M., Kostikov, R. R., Kisin, A. V. & Babaev, N. M. (1991). Zh. Org. Khim. 27, 1434-1440 (in Russian); Chem. Abstr. (1992), 116, 807.
- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.
- Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
- Cukurovalı, A., Yılmaz, I., Özmen, H. & Ahmedzade, M. (2002). Transition Met. Chem. 27, 171-176.
- Dehmlow, E. V. & Schmidt, S. S. (1990). Liebigs Ann. Chem. pp. 411-414.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Gompper, R. & Christmann, O. (1959). Chem. Ber. 92, 1994-1998.
- Roger, E., Pierre, C. J., Pualette, V., Gerard, G., Chepat, J. P. & Robert, G. (1977). Eur. J. Med. Chem. Chem. Ther. 12, 501-509.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Stoe & Cie (2002). X-AREA (Version 1.18) and X-RED32 (Version 1.04). Stoe & Cie. Darmstadt, Germany.
- Swenson, D. C., Yamamoto, M. & Burton, D. J. (1997). Acta Cryst. C53, 1445-1447
- Yüksektepe, Ç., Saraçoğlu, H., Koca, M., Çukurovalı, A. & Çalışkan, N. (2004). Acta Cryst. C60, o509-o510.